

CALICUT UNIVERSITY – FOUR-YEAR UNDER GRADUATE PROGRAMME (CU-FYUGP)

BSc CHEMISTRY

Programme	B. Sc. Chemistry						
Course Title	ENVIRONMENTAL CHEMISTRY						
Type of Course	MDC						
Semester	Ι						
Academic	100-199						
Level							
Course Details	Credit	Lecture per	Tutorial	Practical	Total Hours		
		week	per week	per week			
	3	3	-	-	45		
Pre-requisites	What is Environment.						
	Basic idea of environmental pollution.						
Course	This course ensures that the students acquire a profound knowledge and						
Summary	understanding on environmental pollution and the necessity of controlling						
	environmental pollution.						

Course Outcomes (CO):

CO	CO Statement	Cognitive Level*	Knowledge Category#	Evaluation Tools used
CO1	Acquire the knowledge on ecosystem.	U	C	Instructor- created exams / Quiz
CO2	Recall the technical/scientific terms involved in pollution.	U	С	Instructor- created exams / Quiz
CO3	Recognize different types of toxic substances that cause environmental pollution.	U	С	Instructor- created exams / Assignment
CO4	Understand the effects of environmental pollution.	U	С	Seminar Presentation / Viva
CO5	Understand various pollution control measures.	U	С	Instructor- created exams / Quiz
CO6	Discuss and report local and global environmental issues based on the knowledge gained throughout the course.	Ар	Р	Group discussion and Seminar presentation/Viv a

* - Remember (R), Understand (U), Apply (Ap), Analyse (An), Evaluate (E), Create (C)
- Factual Knowledge(F) Conceptual Knowledge (C) Procedural Knowledge (P)
Metacognitive Knowledge (M)

Detailed Syllabus:

Module	Unit	Hrs	Mark	
		Introduction to Environmental Chemistry	9	18
Ι	1	Environmental segments-Atmosphere, Hydrosphere, Lithosphere,	2	
		Biosphere		
	2	Interaction between different environmental spheres Concept of		
		ecosystem, abiotic and biotic components		
	3	Composition of Air, Water and Soil	2	
	4	Environmental pollution – Concepts and definition – Pollutant,	1	
		contaminant, receptor and sink		
	5	Classification of pollutants - Global, regional, local, persistent and non-	1	
		persistent pollutants.		
	6	Types of pollution	1	
II		Air Pollution	9	18
	7	Tropospheric pollution – Gaseous air pollutants – Hydrocarbons,	2	
		oxides of sulphur, nitrogen and carbon (Elementary idea only)		
	8	Global warming, green house effect, acid rain	1	
	9	Particulates - Smog: London smog and photochemical smog -	2	
	10	stratospheric pollution - depletion of ozone layer, chlorofluorocarbons -	2	
		Automobile pollution.		
	11	Control of air pollution	2	
III		Water Pollution	10	20
	12		1	20
	12	Impurities in water – cause of pollution – natural and anthropogenic –	1	
		Marine water pollution – Underground water pollution.		
	13		2	
		Source of water pollution – Industrial waste, Municipal waste,		
		Agricultural waste, Radioactive waste, Petroleum, Pharmaceutical,		
		heavy metal, pesticides, soaps and detergents.		
			-	
	14	Types of water pollutants: Biological agents, physical agents and	2	
		chemical agents – Eutrophication- biomagnification and		
	1.5	bloaccumulation.		
	15	Water quality parameters: DO, BOD, COD, alkalianity, hardness,	3	
		chloride, fluoride and nitrate. Toxic metals in water and their effects:		
		Cadmium, lead and oil pollution in water.		
		· •		
	16		2	
		water pollution control methods		

IV	Soil, Thermal, and Radioactive Pollutions					
	18	Soil pollution: Sources by industrial and urban wastes. Non-degradable, degradable and biodegradable wastes. Hazardous waste.	2			
	19	Pollution due to plastics, pesticides, biomedical waste and <i>e-waste</i> (source, effects and control measures) – Control of soil pollution - Solid waste Management – Open dumping, Landfilling, Incineration, Reuse, reclamation, recycle, composting.	3			
	20	Thermal pollution – definition, sources, harmful effects and prevention.	1			
	21	Radioactive pollution (source, effects and control measures) – Hiroshima, Nagasaki and Chernobyl accidents (brief study).	2			
V		Open Ended Module: Environmental issues	9			
	1	Environment and society Pollution case studies: Chernobyl disaster, Bhopal tragedy, Endosulfan disaster in Kerala (brief study) etc.				

References

- 1. A. K. De, *Environmental Chemistry*, 6thEdn., New Age International.
- 2. A. K. Ahluwalia, *Environmental Chemistry*, The Energy and Resources Institute, 2017.
- 3. Balram Pani, Textbook of Environmental Chemistry, I. K. International Pvt Ltd, 2010.

4. S.K. Banergy, *Environmental Chemistry*, 2nd Edn., Prentice-Hall of India Pvt. Ltd., New Delhi, 2005.

5. V.N. Bashkin, *Environmental Chemistry: Asian Lessons*, Springer Science & Business Media, 2003.

6. S.E. Manahan, *Environmental Chemistry*, 8th Edn., CRC Press, Florida, 2004.

7.Balram Pani, Textbook of Environmental Chemistry, I. K. International Pvt Ltd, 2010.

8. J. M. H. Selendy, *Water and Sanitation-Related Diseases and the Changing Environment*, John Wiley & Sons, Inc.

9. P. K. Goel, Water Pollution: Causes, Effects and Control, New Age International, 2006.

10. V. N. Bashkin, *Environmental Chemistry: Asian Lessons*, Springer Science & Business Media, 2003.

11.Pallavi Saxena, Vaishali Naik, Air Pollution: Sources, Impacts and Controls, CAB International, 2018.

- 12. Gabi Mocatta(2015) Environmental Journalism, Deakin University Open School of Journalism.
- 13. D. S. Poornananda (2022), Environmental Journalism: Reporting on Environmental Concerns and Climate Change in India, SAGE Publishing India'
- 14. Frome, Michael. (1998) Green Ink: An Introduction to Environmental Journalism. Salt Lake City: University of Utah Press.
- 15. Sachsman, D. B. & Valenti, J. M. (2020). Routledge handbook of environmental *journalism*.New York, NY: Routledge.
- 16. Blum, D., Henig, R., Knudson, M., (2005). "<u>A Field Guide for Science Writers</u>." Oxford University Press; 2nd edition.
- 17. Hansen, Anders. (2010) Environment, Media and Communication. London: Routledge

P					- 05	•							
	PSO 1	PSO 2	PSO 3	PSO4	PS O5	PSO 6	PO1	PO2	PO3	PO4	PO5	PO6	PO7
CO 1	1	-	-	-	1	1	1			2	1		
CO 2	1		-	-	1	1	1			1	1	1	1
CO 3	-	-		1	2	2	1			2	2	1	
CO 4	-	-			1	2	1			1	1	1	1
CO 5	-		-	1	2	2	1			1		1	1
CO 6	-	-	-	1	2	2	1			1	1	1	1

Mapping of COs with PSOs and POs :

Correlation Levels:

Level	Correlation	
-	Nil	
1	Slightly / Low	
2	Moderate /	
	Medium	
3	Substantial /	
	High	

Assessment Rubrics:

- Quiz / Assignment/ Quiz/ Discussion / Seminar
- Midterm Exam
- Programming Assignments (20%)
- Final Exam (70%)

	Internal Exam	Assignm ent/viva	Quiz/seminar/ Goupdiscussio n	End Semester Examinations
CO 1	\checkmark		\checkmark	\checkmark
CO 2	\checkmark		\checkmark	\checkmark
CO 3	\checkmark	\checkmark		\checkmark
CO 4		\checkmark	\checkmark	\checkmark
CO 5	\checkmark		\checkmark	\checkmark
CO 6		\checkmark	\checkmark	

Mapping of COs to Assessment Rubrics :